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analysis of fact
ors...Metallic
Silver as an**

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cores from the**

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precious metals.

**A new method of
ore appraisal is
introduced which
estimates grade
and tonnage of
ore by
calculating
convex hulls
(smallest convex
polygons) for
sample points
with assay values
above specified**

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cutoff grades.

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Leong.
Submitted... [at]
Columbia
University.
[Preface by**

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Each Month". This complete summary of the ideas from Yanik Silver's book "Moonlighting on the Internet" shows how nowadays it is entirely feasible to "moonlight on the internet" and make some extra money

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every month while holding down a full-time job elsewhere. By working in your spare time, you can build an online business which generates whatever level of regular income you want. In his book, the author states that to key is

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to have a strategy to make the right things happen, rather than just sitting back and hoping for the best.

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value of this
summary: □ Save
time □ Understand
key concepts □
Expand your
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more, read
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time to gain some

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extra income.
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this Campaign
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Manual of
Qualitative
Analysis It is***

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well known that silver chloride is insoluble in water, and the presence of the sodium nitrate in the resulting liquid does not seem to affect the solubility of the precipitate. If

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*the Silver
chloride be
separated from
the solution
and divided
into two parts,
and one part
treated with
dilute nitric
acid and the
other with
ammonium
hydrate, it is*

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found that the first part does not dissolve, while the second part does dissolve. It is fair to assume, then, that if nitric acid is present with the silver nitrate when the sodium

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chloride is added, the precipitation of the silver chloride is not obstructed but if ammonium hydrate is present with the silver nitrate when the sodium chloride is

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added, the expected precipitate of silver chloride will not come down, though silver is present in the solution. This may easily be verified by experiment, and shows the

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***importance of
keeping track
of everything
that may be
present in a
solution under
examination.***

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Silver from the
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Future of the White
Metal But an
analysis of the silver
market of the last
few years. And of the
factors that have
been at play, will
demonstrate that
Cobalt production
has been a negligible
quantity. About the
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**vast majority of
imperfections
successfully; any
imperfections that
remain are
intentionally left to
preserve the state of
such historical
works.**

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and Gold Solutions
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factors...**

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for Gold and Silver
Silver Economic
Analysis**

**Systems Analysis
and Design**

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This paper considers the directional predictability of daily returns for both gold and silver. These two metals have had a long history behaving sometimes as complements and

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other times as substitutes. We use daily data from June of 2008 through February of 2015. The last two years were removed as a set for validation of the model and the remainder,

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***almost 5 years,
was used as
training. A cluster
analysis yields
six important
clusters. An
evaluation of
these clusters
leads to the
formation of
three strategies
for directional***

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***predictions -- up
or down -- for
both gold and
silver returns.
The results of
this analysis
suggest that each
strategy has its
own advantages:
the first strategy
suggests that
gold returns can***

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***be predicted
better than those
of silver; the
second strategy
shows that
predicting up for
gold also means
predicting down
for silver and the
final strategy
confirms that
predicting up for***

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***silver also
validates***

***predicting down
for gold.***

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*Excerpt from
The Analysis
of Cyanide
Silver-Plating
Solutions In
former years
the baths were
made from*

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*silver
chloride, and
hence*

*contained much
alkali
chloride. Some
chloride is
present in all
commercial
baths as a
result of
impurities in*

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*the chemicals
or the water
used. The
baths are
likely to
contain iron,
present as
ferrocyanide.
This enters
from corrosion
of the steel
tanks, pipes,*

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*or hangers,
especially if
much chloride
is present.*

*Small amounts
of copper may
be introduced
from the
preceding
strike
solutions, to
which copper*

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*is added, or
from corrosion
of the copper
tank bars.*

*Mercury may be
introduced,
especially
into the
strike
solution, if
mercury dips
are used prior*

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plating. As
mercury dips
are now seldom
used, such
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