

## Chemistry Atoms First By Julia Burdge

This print companion to MindTap General Chemistry: Atoms First presents the narrative, figures, tables and example problems—but no graded problems or assessments. Students must use MindTap to complete the interactive activities, exercises, and assignments. The atoms first organization introduces students to atoms and molecules earlier and delays math-intensive problem-solving to later in the semester. This gives students a stronger conceptual framework to help them succeed in the course. In addition, the narrative provides greater emphasis on the historical development of the atomic nature of matter and atomic structure. Important Notice: Media content referenced within the product description or the product text may not be available in the ebook version.

Helping you focus on mastering the quantitative skills and conceptual knowledge you need to get a true understanding of chemistry, this text continues the tradition of relevance that makes it so effective. Now including MasteringChemistry, the online homework, tutorial, and assessment product with a demonstrated record of helping students quickly master concepts, this edition includes new opportunities for you to practice key concepts. MasteringChemistry provides seamless synergy with the text to create a dynamic learning program that enables you to learn both in and out of the classroom.

An Introduction to Chemistry is intended for use in beginning chemistry courses that have no chemistry prerequisite. The text was written for students who want to prepare themselves for general college chemistry, for students seeking to satisfy a science requirement for graduation, and for students in health-related or other programs that require a one-semester introduction to general chemistry. No matter what a reader's goals are, this book will help them to learn the basics of chemistry.

Steve and Susan Zumdahl's texts focus on helping students build critical -thinking skills through the process of becoming independent problem-solvers. They help students learn to think like chemists so they can apply the problem solving process to all aspects of their lives. In this Second Edition of CHEMISTRY: AN ATOMS FIRST APPROACH, the Zumdahls use a meaningful approach that begins with the atom and proceeds through the concept of molecules, structure, and bonding, to more complex materials and their properties. Because this approach differs from what most students have experienced in high school courses, it encourages them to focus on conceptual learning early in the course, rather than relying on memorization and a plug and chug method of problem solving that even the best students can fall back on when confronted with familiar material. The atoms first organization provides an opportunity for students to use the tools of critical thinkers: to ask questions, to apply rules and models, and to evaluate outcomes. Important Notice: Media content referenced within the product description or the product text may not be available in the ebook version.

A Worked Examples Approach

Atoms First 2e

General Chemistry

An Atoms-Focused Approach

"General Chemistry: Atoms First," Second Edition starts from the building blocks of chemistry, the atom, allowing the authors to tell a cohesive story that progresses logically through molecules and compounds to help students intuitively follow complex concepts more logically. This unified thread of ideas helps students build a better foundation and ultimately gain a deeper understanding of chemical concepts. Students can more easily understand the microscopic-to-macroscopic connections between unobservable atoms and the observable behavior of matter in daily life, and are brought immediately into real chemistryinstead of being forced to memorize facts. Reflecting a true atoms first perspective, the Second Edition features experienced atoms-first authors, incorporates recommendations from a panel of atoms-first experts, and follows historical beliefs in teaching chemistry concepts based and real experimental data first. This approach distinguishes this text in the market based whereby other authors teach theory first, followed by experimental data.

Bishop's text shows students how to break the material of preparatory chemistry down and master it. The system of objectives tells the students exactly what they must learn in each chapter and where to find it.

The Atoms First approach provides a consistent and logical method for teaching general chemistry. This approach starts with the fundamental building block of matter, the atom, and uses it as the stepping-stone to understanding more complex chemistry topics. Once mastery of the nature of atoms and electrons is achieved, the formation and properties of compounds are developed. Only after the study of matter and the atom will students have sufficient background to fully engage in topics such as stoichiometry, kinetics, equilibrium, and thermodynamics. Thus, the Atoms First approach empowers instructors to present the most complete and compelling story of general chemistry. Far from a simple re-ordering of topics, this is a book that will truly meet the needs of the growing atoms-first market. The fourth edition continues to build on the innovative success of the previous three editions. Changes to this edition include specific refinements intended to augment the student-centered pedagogical features that continue to make this book effective and popular both with professors, and with their students.

From its very origin, Introductory Chemistry: An Atoms First Approach has been developed and written using an atoms first approach specific to introductory chemistry. It is not a pared down version of a general chemistry text, but carefully crafted with the introductory chemistry student in mind.The ordering of topics facilitates the conceptual development of chemistry for the novice, rather than the historical development that has been used traditionally. Its language and style are student friendly and conversational and the importance and wonder of chemistry in everyday life are emphasized at every opportunity. Continuing in the Burdge tradition, this text employs an outstanding art program, a consistent problem-solving approach, interesting applications woven throughout the chapters and a wide range of end-of-chapter problems.

General Chemistry: Atoms First

Calculus

Part 2: Atoms First

Loose Leaf Version for Chemistry: Atoms First

ALERT: Before you purchase, check with your instructor or review your course syllabus to ensure that you select the correct ISBN. Several versions of Pearson's MyLab & Mastering products exist for each title, including customized versions for individual schools, and registrations are not transferable. In addition, you may need a CourseID, provided by your instructor, to register for and use Pearson's MyLab & Mastering products. Packages Access codes for Pearson's MyLab & Mastering products may not be included when purchasing or renting from companies other than Pearson; check with the seller before completing your purchase. Used or rental books If you rent or purchase a used book with an access code, the access code may have been redeemed previously and you may have to purchase a new access code. Access codes Access codes that are purchased from sellers other than Pearson carry a higher risk of being either the wrong ISBN or a previously redeemed code. Check with the seller prior to purchase. xxxxxxxxxxxx Carrying through an atoms-first approach from the first four editions, and helping you focus on mastering the quantitative skills and conceptual knowledge you need to get a true understanding of chemistry, Russo and Silver's Introductory Chemistry, Fifth Edition continues the tradition of relevance that makes it so effective. Now including MasteringChemistry®, the leading online homework, tutorial, and assessment product with a demonstrated record of helping students quickly master concepts, this Fifth Edition includes new opportunities for you to practice key concepts. MasteringChemistry provides seamless synergy with the text to create a dynamic learning program that enables you to learn both in and out of the classroom. With Russo and Silver's Introductory Chemistry, Fifth Edition and MasteringChemistry, you get a complete teaching and learning program that gives you critical tools for ensuring a successful introduction to chemistry, including: An atoms-first approach to chemistry: Through an atoms-first approach used effectively in the previous four editions, you begin to learn starting from the building blocks of matter and progress to understanding complex concepts from a logical point of view and with a deep understanding. Personalized, interactive learning for achieving proficiency of the concepts with MasteringChemistry: Self-paced tutorials guide you through the text's most challenging topics; provide immediate, specific feedback and reinforcement; and present varied content to keep you engaged and on track. An emphasis on core concepts for solving quantitative and qualitative problems: Get a true understanding of introductory chemistry by using material that presents problem solving and comprehension as complimentary skills, rather than encouraging rote memorization. Features that demonstrate how relevant chemistry concepts are in students' lives: A number of outstanding features that show chemistry as a fascinating science.

Laboratory Manual to Accompany Chemistry: Atoms First by Gregg Dieckmann and John Sibert from the University of Texas at Dallas. This laboratory manual presents a lab curriculum that is organised around an atoms-first approach to general chemistry. The philosophy behind this manual is to (1) provide engaging experiments that tap into student curiosity, (2) emphasize topics that students find challenging in the general chemistry lecture course, and (3) create a laboratory environment that encourages students to "solve puzzles" or "play" with course content and not just "follow recipes." The laboratory manual represents a terrific opportunity to get students turned on to science while creating an environment that connects the relevance of the experiments to a greater understanding of their world. This manual has been written to provide instructors with tools that engage students, while providing important connections to the material covered in an atoms-first lecture course.

Chemistry: Atoms First 2e is a peer-reviewed, openly licensed introductory textbook produced through a collaborative publishing partnership between OpenStax and the University of Connecticut and UConn Undergraduate Student Government Association. This text is an atoms-first adaptation of OpenStax Chemistry 2e. The intention of "atoms-first" involves a few basic principles: first, it introduces atomic and molecular structure much earlier than the traditional approach, and it threads these themes through subsequent chapters. This approach may be chosen as a way to delay the introduction of material such as stoichiometry that students traditionally find abstract and difficult, thereby allowing students time to acclimate their study skills to chemistry. Additionally, it gives students a basis for understanding the application of quantitative principles to the chemistry that underlies the entire course. It also aims to center the study of chemistry on the atomic foundation that many will expand upon in a later course covering organic chemistry, easing that transition when the time arrives. The second edition has been revised to incorporate clearer, more current, and more dynamic explanations, while maintaining the same organization as the first edition. Substantial improvements have been made in the figures, illustrations, and example exercises that support the text narrative.

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An Atoms First Approach 2e

Chemistry of Free Atoms and Particles

Chemistry: An Atoms First Approach

Fundamentals of Chemistry

From its very origin, Introductory Chemistry: An Atoms First Approach by Julia Burdge and Michelle Driessen has been developed and written using an atoms-first approach specific to introductory chemistry. It is not a pared down version of a general chemistry text, but carefully crafted with the introductory-chemistry student in mind. The ordering of topics facilitates the conceptual development of chemistry for the novice, rather than the historical development that has been used traditionally. Its language and style are student-friendly and conversational; and the importance and wonder of chemistry in everyday life are emphasized at every opportunity. Continuing in the Burdge tradition, this text employs an outstanding art program, a consistent problem-solving approach, interesting applications woven throughout the chapters, and a wide range of end-of-chapter problems.

"Chemistry is designed for the two-semester general chemistry course. For many students, this course provides the foundation to a career in chemistry, while for others, this may be their only college-level science course. As such, this textbook provides an important opportunity for students to learn the core concepts of chemistry and understand how those concepts apply to their lives and the world around them. The text has been developed to meet the scope and sequence of most general chemistry courses. At the same time, the book includes a number of innovative features designed to enhance student learning. A strength of Chemistry is that instructors can customize the book, adapting it to the approach that works best in their classroom."--Openstax College website.

This is part two of two for Chemistry: Atoms First by OpenStax. This book covers chapters 11-21. Chemistry: Atoms First is a peer-reviewed, openly licensed introductory textbook produced through a collaborative publishing partnership between OpenStax and the University of Connecticut and UConn Undergraduate Student Government Association. This title is an adaptation of the OpenStax Chemistry text and covers scope and sequence requirements of the two-semester general chemistry course. Reordered to fit an atoms first approach, this title introduces atomic and molecular structure much earlier than the traditional approach, delaying the introduction of more abstract material so students have time to acclimate to the study of chemistry. Chemistry: Atoms First also provides a basis for understanding the application of quantitative principles to the chemistry that underlies the entire course. The images in this textbook are grayscale.

Electrons, Atoms, and Molecules in Inorganic Chemistry: A Worked Examples Approach builds from fundamental units into molecules, to provide the reader with a full understanding of inorganic chemistry concepts through worked examples and full color illustrations. The book uniquely discusses failures as well as research success stories. Worked problems include a variety of types of chemical and physical data, illustrating the interdependence of issues. This text contains a bibliography providing access to important review articles and papers of relevance, as well as summaries of leading articles and reviews at the end of each chapter so interested readers can readily consult the original literature. Suitable as a professional reference for researchers in a variety of fields, as well as course use and self-study. The book offers valuable information to fill an important gap in the field. Incorporates questions and answers to assist readers in understanding a variety of problem types Includes detailed explanations and developed practical approaches for solving real chemical problems Includes a range of example levels, from classic and simple for basic concepts to complex questions for more sophisticated topics Covers the full range of topics in inorganic chemistry: electrons and wave-particle duality, electrons in atoms, chemical binding, molecular symmetry, theories of bonding, valence bond theory, VSEPR theory, orbital hybridization, molecular orbital theory, crystal field theory, ligand field theory, electronic spectroscopy, vibrational and rotational spectroscopy

Introductory Chemistry: An Atoms First Approach

A New System of Chemical Philosophy . . .

An Introduction to Chemistry - Atoms First

AP Chemistry

The atoms first approach provides a consistent and logical method for teaching general chemistry. This approach starts with the fundamental building block of matter, the atom, and uses it as the stepping stone to understanding more complex chemistry topics. Once mastery of the nature of atoms and electrons is achieved, the formation and properties of compounds are developed. Only after the study of matter and the atom will students have sufficient background to fully engage in topics such as stoichiometry, kinetics, equilibrium, and thermodynamics. Thus, the Atoms First method empowers instructors to present the most complete and compelling story of general chemistry. Far from a simple re-ordering of topics, this is a book that will truly meet the needs of the growing atoms-first market.

"Atoms First seems to be the flavor of the year in chemistry textbooks, but many of them seem to be little more than rearrangement of the chapters. It takes a master like McQuarrie to go back to the drawing board and create a logical development from smallest to largest that makes sense to students."--Hal Harris, University of Missouri-St. Louis "McQuarrie's book is extremely well written, the order of topics is logical, and it does a great job with both introductory material and more advanced concepts. Students of all skill levels will be able to learn from this book."--Mark Kearley, Florida State University This new fourth edition of General Chemistry takes an atoms-first approach from beginning to end. In the tradition of McQuarrie's many previous works, it promises to be another ground-breaking text. This superb new book combines the clear writing and wonderful problems that have made McQuarrie famous among chemistry professors and students worldwide. Presented in an elegant design with all-new illustrations, it is available in a soft-cover edition to offer professors a fresh choice at an outstanding value. Student supplements include an online series of descriptive chemistry Interchapters, a Student Solutions Manual, and an optional state-of-the-art Online Homework program. For adopting professors, an instructor's Manual and a CD of the art are also available.

Most people remember chemistry from their school days as largely incomprehensible, a subject that was fact-rich but understanding-poor, smelly, and so far removed from the real world of events and pleasures that there seemed little point, except for the most introverted, in coming to terms with its grubby concepts, spells, recipes, and rules. Peter Atkins wants to change all that. In this Very Short Introduction to Chemistry, he encourages us to look at chemistry anew, through a chemist's eyes, in order to understand its central concepts and to see how it contributes not only towards our material comfort, but also to human culture. Atkins shows how chemistry provides the infrastructure of our world, through the chemical industry, the fuels of heating, power generation, and transport, as well as the fabrics of our clothing and furnishings. By considering the remarkable achievements that chemistry has made, and examining its place between both physics and biology, Atkins presents a fascinating, clear, and rigorous exploration of the world of chemistry - its structure, core concepts, and exciting contributions to new cutting-edge technologies. ABOUT THE SERIES: The Very Short Introductions series from Oxford University Press contains hundreds of titles in almost every subject area. These pocket-sized books are the perfect way to get ahead in a new subject quickly. Our expert authors combine facts, analysis, perspective, new ideas, and enthusiasm to make interesting and challenging topics highly readable.

Packed with the information, examples and problems you need to learn to think like a chemist, CHEMISTRY: AN ATOMS FIRST APPROACH, Third Edition is designed to help you become an independent problem-solver. The text begins with coverage of the atom and proceeds through the concept of molecules, structure and bonding. This approach, different from your high school course, will help you become an adept critical thinker and a strong problem-solver -- skills that will be useful to you in any career. Important Notice: Media content referenced within the product description or the product text may not be available in the ebook version.

Student Workbook and Selected Solutions for Introductory Chemistry

Loose Leaf for Chemistry: Atoms First

Introductory Chemistry

Chemistry 2e

*Steve and Susan Zumdahl's texts focus on helping students build critical thinking skills through the process of becoming independent problem-solvers. They help students learn to think like a chemists so they can apply the problem solving process to all aspects of their lives. In CHEMISTRY: AN ATOMS FIRST APPROACH, the Zumdahls use a meaningful approach that begins with the atom and proceeds through the concept of molecules, structure, and bonding, to more complex materials and their properties. Because this approach differs from what most students have experienced in high school courses, it encourages them to focus on conceptual learning early in the course, rather than relying on memorization and a plug and chug method of problem solving that even the best students can fall back on when confronted with familiar material. The atoms first organization provides an opportunity for students to use the tools of critical thinkers: to ask questions, to apply rules and models and to evaluate outcomes. Important Notice: Media content referenced within the product description or the product text may not be available in the ebook version.*

*Chemistry: General Chemistry:Atoms First/Prentice Hall*

*This is part one of two for Chemistry by OpenStax. This book covers chapters 1-11. Chemistry is designed for the two-semester general chemistry course. For many students, this course provides the foundation to a career in chemistry, while for others, this may be their only college-level science course. As such, this textbook provides an important opportunity for students to learn the core concepts of chemistry and understand how those concepts apply to their lives and the world around them. The text has been developed to meet the scope and sequence of most general chemistry courses. At the same time, the book includes a number of innovative features designed to enhance student learning. A strength of Chemistry is that instructors can customize the book, adapting it to the approach that works best in their classroom. The images in this textbook are grayscale.*

*Gilbert Strang's clear, direct style and detailed, intensive explanations make this textbook ideal as both a course companion and for self-study. Single variable and multivariable calculus are covered in depth. Key examples of the application of calculus to areas such as physics, engineering and economics are included in order to enhance students' understanding. New to the third edition is a chapter on the 'Highlights of calculus', which accompanies the popular video lectures by the author on MIT's OpenCourseWare. These can be accessed from math.mit.edu/~gs.*

*The Central Science*

*Atoms First*

*Lab Manual for Chemistry: Atoms First*

*Laboratory Manual for General Chemistry*

The authors, who have more than two decades of combined experience teaching an atoms-first course, have gone beyond reorganizing the topics. They emphasize the particulate nature of matter throughout the book in the text, art, and problems, while placing the chemistry in a biological, environmental, or geological context. The authors use a consistent problem-solving model and provide students with ample opportunities to practice.

Interactive General Chemistry meets students where they are...with a general chemistry program designed for the way students learn. Achieve provides a new platform for Interactive General Chemistry, thoughtfully developed to engage students for better outcomes. Powerful data and analytics provide instructors with actionable insights on a platform that allows flexibility to align with a broad variety of teaching and learning styles and the exciting Interactive General Chemistry program! Whether a student's learning path starts with problem solving or with reading, Interactive General Chemistry delivers the learning experience he or she needs to succeed in general chemistry. Built from the ground up as a digital learning program, Interactive General Chemistry combines the Sapling Learning homework platform with a robust e-book with seamlessly embedded, multimedia-rich learning resources. This flexible learning environment helps students effectively and efficiently tackle chemistry concepts and problem solving. Student-centered development in addition to Macmillan's standard rigorous peer review process, student involvement was critical to the development and design of Interactive General Chemistry. Using extensive research on student study behavior and data collection on the resources and tools that most effectively promote understanding, we crafted this complete course solution to intentionally embrace the way that students learn. Digital-first experience Interactive General Chemistry was built from the ground up to take full advantage of the digital learning environment. High-quality multimedia resources--including Sapling Interactives, pHET simulations, and new whiteboard videos by Tyler DeWitt--are seamlessly integrated into a streamlined, uncluttered e-book. Embedded links provide easy and efficient navigation, enabling students to link to review material and definitions as needed. Problems drive purposeful study Our research into students' study behavior showed that students learn best by doing--so with Interactive General Chemistry, homework problems are designed to be a front door for learning. Expanding upon the acclaimed Sapling homework--where every problem contains hints, targeted feedback, and detailed step-by-step solutions--embedded resources link problems directly to the multimedia-rich e-book, providing just-in-time support at the section and chapter level.

General Chemistry: Atoms First, Second Edition starts from the building blocks of chemistry, the atom, allowing the authors to tell a cohesive story that progresses logically through molecules and compounds to help students intuitively follow complex concepts more logically. This unified thread of ideas helps students build a better foundation and ultimately gain a deeper understanding of chemical concepts. Students can more easily understand the microscopic-to-macroscopic connections between unobservable atoms and the observable behavior of matter in daily life, and are brought immediately into real chemistry-instead of being forced to memorize facts. Reflecting a true atoms first perspective, the Second Edition features experienced atoms-first authors, incorporates recommendations from a panel of atoms-first experts, and follows historical beliefs in teaching chemistry concepts based and real experimental data first. This approach distinguishes this text in the market based whereby other authors teach theory first, followed by experimental data.

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Chemistry

Student Solutions Manual for Zumdahl/DeCoste's Chemical Principles, 7th

Study Guide for an Introduction to Chemistry - Atoms First

Features examples from each chapter, learning objectives, review of key concepts from the text, and additional problems for student practice. Also, the workbook provides comprehensive answers and explanations to selected end-of-chapter problems from the text. Provides over 200 worked examples and more than 550 practice problems and quiz questions to help students develop and practice their problem-solving skills.

Chemistry of Free Atoms and Particles covers the chemistry of metal atoms and metallic molecules or fragments. This book contains 10 chapters that are organized on the basis of the Periodic Chart. Each group of elements is separated into a discussion of first the free atoms, followed by a discussion of reactive molecular forms of metal halides, oxides, and sulfides. These sections are further broken down into subsections on "Occurrence, Properties, and Techniques" followed by "Chemistry". The "Chemistry" sections are further divided into several headings, including abstraction, electron-transfer, oxidative addition, simple orbital mixing, substitution, disproportionation and ligand transfer, and cluster formation processes. This book will be of value to chemistry researchers, teachers, and students.

"Chemistry: Atoms First is a peer-reviewed, openly licensed introductory textbook produced through a collaborative publishing partnership between OpenStax and the University of Connecticut and UConn Undergraduate Student Government Association. This title is an adaptation of the OpenStax Chemistry text and covers scope and sequence requirements of the two-semester general chemistry course. Reordered to fit an atoms first approach, this title introduces atomic and molecular structure much earlier than the traditional approach, delaying the introduction of more abstract material so students have time to acclimate to the study of chemistry. Chemistry: Atoms First also provides a basis for understanding the application of quantitative principles to the chemistry that underlies the entire course."--Open Textbook Library.

Fundamentals of Chemistry, Fourth Edition covers the fundamentals of chemistry. The book describes the formation of ionic and covalent bonds; the Lewis theory of bonding; resonance; and the shape of molecules. The book then discusses the theory and some applications of the four kinds of spectroscopy: ultraviolet, infrared, nuclear (proton) magnetic resonance, and mass. Topics that combine environmental significance with descriptive chemistry, including atmospheric pollution from automobile exhaust; the metallurgy of iron and aluminum; corrosion; reactions involving ozone in the upper atmosphere; and the methods of controlling the pollution of air and water, are also considered. Chemists and students taking courses related to chemistry and environmental chemistry will find the book invaluable.

An Introduction to Chemistry

Electrons, Atoms, and Molecules in Inorganic Chemistry

Second Edition

Interactive Chemistry Achieve, 1-term Access Code

By Stephanie Dillon with contributions from Sandra Chimon Peszek, DePaul University Laboratory Manual for General Chemistry: Atoms First, Second Edition is organized using the atoms first approach and is written to correspond with the Second Edition of General Chemistry: Atoms First by McMurry/Fay. This manual contains twenty-four experiments with a focus on real world applications, following an intuitive logic progressing from the simplest building blocks to successively more complex concepts. Each experiment covers one or more topics discussed within a chapter of the textbook to help students understand the underlying concepts covered in the lecture course. Additionally, each experiment contains a set of pre-laboratory questions (also assignable in MasteringChemistry®), an introduction, a background section explaining concepts that each student is expected to master for a full understanding of the experimental results, a step-by-step procedure (including safety information), and a report section featuring post-laboratory questions. Note: This is the standalone book (Laboratory Manual for General Chemistry: Atoms First, Second Edition) if you want the book/access card order the ISBN below: You must have the Instructor ID to access MasteringChemistry. 0321913329 / 9780321913326 General Chemistry: Atoms First Plus MasteringChemistry with eText -- Access Card Package & Laboratory Manual for General Chemistry: Atoms First Package\* Package consists of: 032180483X / 9780321804839 General Chemistry: Atoms First Plus MasteringChemistry with eText -- Access Card Package 0321813375 / 9780321813374 Laboratory Manual for General Chemistry: Atoms First

Chemistry: A Very Short Introduction

An Atoms First Approach