

Introduction To Phase Equilibria In Ceramics

Introduction to Phase Equilibria in Ceramic SystemsRoutledge

The book begins with an overview of the phase diagrams of fluid mixtures (fluid – liquid, gas, or supercritical state), which can show an astonishing variety when elevated pressures are taken into account; phenomena like retrograde condensation (single and double) and azeotropy (normal and double) are discussed. It then gives an introduction into the relevant thermodynamic equations for fluid mixtures, including some that are rarely found in modern textbooks, and shows how they can be used to compute phase diagrams and related properties. This chapter gives a consistent and axiomatic approach to fluid thermodynamics; it avoids using activity coefficients. Further chapters are dedicated to solid-fluid phase equilibria and global phase diagrams (systematic search for phase diagram classes). The appendix contains numerical algorithms needed for the computations. The book thus enables the reader to create or improve computer programs for the calculation of fluid phase diagrams. introduces phase diagram classes, how to recognize them and identify their characteristic features presents rational nomenclature of binary fluid phase diagrams includes problems and solutions for self-testing, exercises or seminars

ThisEncyclopedia begins with an introduction summarizing isotope and content. Glassmaking; Structure of Glass, GlassPhysics,Transport Properties, Chemistry of Glass, Glass and Light,Inorganic Glass Families, Organic Glasses, Glass and theEnvironment, Historical and Economical Aspect of Glassmaking,History of Glass, Glass and Art, and outline/new developments and uses as presented by the best known people in thefield (C.A. Angell, for example). Sections and chapters arearranged in a logical order to ensure overall consistency and avoidsuseless repetitions. All sections are introduced by a briefintroduction and attractive illustration. Newly investigatedtopics will be addressed, with the goal of ensuring that thisEncyclopedia remains a reference work for years to come.

This book provides a comprehensive overview of ionic liquid based separation techniques. The glimpse of thermodynamic predictive models along with global optimization techniques will help readers understand the separation techniques at molecular and macroscopic levels. Experimental and characterization techniques are coupled with model based predictions so as to provide multicomponent data for the scientific community. The models will focus more on the a-priori based predictions which gives higher emphasis on hydrogen-bonded systems. Particle Swarm Optimization (PSO) technique will also eventually help the readers to apply optimization technique to an extraction process. The overriding goal of this work is to provide pathways for leading engineers and researchers toward a clear understanding and firm grasp of the phase equilibria of Ionic Liquid systems.

High-Pressure Fluid Phase Equilibria Phase Diagrams

Encyclopedia of Glass Science, Technology, History, and Culture Two Volume Set Methods for Phase Diagram Determination

A Comprehensive Reference Source on the Chemistry of the Earth

In this chapter, the basic methodologies of phase equilibrium engineering are introduced through the systematic analysis of several case studies. Some of the thermodynamic tools that have been presented in the previous chapters are applied to illustrate how the phase and conceptual process design of complex engineering problems can be tackled from a phase equilibrium engineering approach. In all the case studies, the first step is to consider in great detail the properties of the process feed, the components, their physical properties, concentrations, and molecular interactions. This information is then used for the selection of thermodynamic models, a suitable technology, pressure, temperature, and compositional operating boundaries. It is shown how the mixture composition and the process goals and specifications determine the process scheme and the unit thermodynamic sensitivity. In addition, the importance of the mixture composition is highlighted in combination with the energy and material balance in the case study for the selection of the desirable natural gas cryogenic technologies. The use of a pressure versus temperature drawing board is used to plot the process trajectory and the mixture phase envelopes from the initial conditions to the key phase engineering design problem. Moreover, the phase design provides also a sound basis for the process initial specification and computer simulation. As another example of phase equilibrium engineering, the heat integration in a complex process is solved by the application of the Gibbs phase rule to the LLV equilibria of a ternary mixture.

Ten years after the debut of the expansive CRC Handbook of Thermodynamic Data of Copolymer Solutions, The CRC Handbook of Phase Equilibria and Thermodynamic Data of Copolymer Solutions updates and expands the world's first comprehensive source of this vital data. Author Christian Wohlfarth, a chemical thermodynamicist specializing in phase equilibria of polymer and copolymer solutions and a respected contributor to the CRC Handbook of Chemistry and Physics, has gathered up-to-the-minute data from more than 500 newly published references. Fully committed to ensuring the reliability of the data, the author included only results with published or personally communicated numerical values. With volumetric, calorimetric, and various phase equilibrium data on more than 450 copolymers and 130 solvents, this handbook furnishes: 150 new vapor-liquid equilibrium datasets 50 new tables containing classical Henry's coefficients 250 new liquid-liquid equilibrium datasets 350 new high-pressure fluid phase equilibrium 70 new PVT-properties datasets 40 new enthalpic datasets Expanded second osmotic virial coefficients data table Carefully organized, clearly presented, and fully referenced, The Handbook of Phase Equilibria and Thermodynamic Data of Copolymer Solutions will prove a cardinal contribution to the open literature and invaluable to anyone working with copolymers.

Phase diagrams are "maps" materials scientists often use to design new materials. They define what compounds and solutions are formed and their respective compositions and amounts when several elements are mixed together under a certain temperature and pressure. This monograph is the most comprehensive reference book on experimental methods for phase diagram determination. It covers a wide range of methods that have been used to determine phase diagrams of metals, ceramics, slags, and hydrides. * Extensive discussion on methodologies of experimental measurements and data assessments * Written by experts around the world, covering both traditional and combinatorial methodologies * A must-read for experimental measurements of phase diagrams

The first volume of this work is organized in three levels, so that the portion and importance of thermodynamics and mathematics increase from level to level. The ground level shows that basics of phase equilibria can be understood without thermodynamics provided the

concept of chemical potential is introduced early. The intermediate level introduces thermodynamics, culminating in the Gibbs energy as the arbiter for equilibrium. At the third level the accent is on binary systems, where one or more phases are solutions of the components. Priority is given throughout to the thermodynamic assessment of experimental data. 200 exercises are included with solutions.

Thermodynamics of the Earth and Planets

The Microstructure-Property Relationship Using Metals as Model Systems

Equilibrium Between Phases of Matter

The Thermodynamics of Phase and Reaction Equilibria

Measurement & Computation

Computational tools allow material scientists to model and analyze increasingly complicated systems to appreciate material behavior. Accurate use and interpretation however, requires a strong understanding of the thermodynamic principles that underpin phase equilibrium, phase formation and state. This fully revised and updated edition covers the fundamentals of thermodynamics, with a view to modern computer applications. The theoretical basis of chemical equilibria and chemical changes is covered with an emphasis on the properties of phase diagrams. Starting with the basic principles, discussion moves to systems involving multiple phases. New chapters cover irreversible thermodynamics, extremum principles, and the thermodynamics of surfaces and interfaces. Theoretical descriptions of equilibrium conditions, the state of systems at equilibrium and the changes as equilibrium is reached, are all demonstrated graphically. With illustrative examples – many computer calculated – and worked examples, this textbook is an valuable resource for advanced undergraduates and graduate students in materials science and engineering.

Studies in Modern Thermodynamics, 3: Phase Theory: The Thermodynamics of Heterogeneous Equilibria focuses on the processes, methodologies, principles, and approaches employed in the determination of the thermodynamics of heterogeneous equilibria. The publication first elaborates on thermodynamic background, phase rule, and general relations for binary equilibria. Discussions focus on linear contributions, G-curves and phase diagram, function changes, Clapeyron's equation, derivation of phase rule, pressure and temperature, heat capacity, enthalpy, and Gibbs energy. The manuscript then examines demixing, isothermal liquid-vapor equilibria, isobaric equilibria between two mixed states, and isobaric equilibria between unmixed solid and mixed liquid state. Topics include phase diagrams, pure solid component with liquid mixture, equilibrium between two ideal states, calculation of phase diagrams, and types of phase diagram. The text ponders on the interpretation of TX phase diagrams and retrograde equilibrium curves, including retrograde solubility, regions of demixing, excess parameters, and eutectic systems. The publication is a valuable reference for researchers wanting to dig deeper into the thermodynamics of heterogeneous equilibria.

Physical chemistry is a compulsory paper offered to all the students of pharmacy. There is a dearth of good books that exclusively cover the syllabi of physical chemistry offered to pharmacy courses. Pharmaceutical Physical Chemistry: Theory and Practices has been designed considering their requirements laid down by AICTE and other premier institutes/universities. Apart from the theory 20 most common laboratory experiments have been included to make this book a unique offering to the students of pharmacy.

After defining complex mixtures, attention is given to the canonical procedure used for the thermodynamics of fluid mixtures: first, we establish a suitable, idealized reference system and then we establish a perturbation (or excess function) which corrects the idealized system for real behavior. For complex mixtures containing identified components (e.g. alcohols, ketones, water) discussion is directed at possible techniques for extending to complex mixtures our conventional experience with reference systems and perturbations for simple mixtures. Possible extensions include generalization of the quasi-chemical approximation (local compositions) and superposition of chemical equilibria (association and solvation) on a physical equation of state. For complex mixtures containing unidentified components (e.g. coal-derived fluids), a possible experimental method is suggested for characterization; conventional procedures can then be used to calculate phase equilibria using the concept of pseudocomponents whose properties are given by the characterization data. Finally, as an alternative to the pseudocomponent method, a brief introduction is given to phase-equilibrium calculations using continuous thermodynamics.

Phase Equilibria for Complex Fluid Mixtures

Phase Equilibria

CALPHAD (Calculation of Phase Diagrams): A Comprehensive Guide

Chapter 6. Phase Equilibrium Engineering Principles

This is a complete and authoritative reference text on an evolving field. Over 200 international scientists have written over 340 separate topics on different aspects of geochemistry including organics, trace elements, isotopes, high and low temperature geochemistry, and ore deposits, to name just a few.

This well-written text is for non-metallurgists and anyone seeking a quick refresher on an essential tool of modern metallurgy. The basic principles, construction, interpretation, and use of alloy phase diagrams are clearly described with ample illustrations for all important liquid and solid reactions. Gas-metal reactions, important in metals processing and in-service corrosion, are also discussed. Get the basics on how phase diagrams help predict and interpret the changes in the structure of alloys.

The rate of growth of stainless steel has outpaced that of other metals and alloys, and by 2010 may surpass aluminum as the second most widely used metal after carbon steel. The 2007 world production of stainless steel was approximately 30,000,000 tons and has nearly doubled in the last ten years. This growth is occurring at the same time that the production of stainless steel continues to become more consolidated. One result of this is a more widespread need to understand stainless steel with fewer resources to provide that information. The concurrent technical evolution in stainless steel and increasing volatility of raw material prices has made it more important for the engineers and designers who use stainless steel to make sound technical judgments about which stainless steels to use and how to use them.

This textbook provides a basic understanding of the formative processes of igneous and metamorphic rock through quantitative applications of simple physical and chemical principles. The book encourages a deeper comprehension of the subject by explaining the petrologic principles rather than simply presenting the student with petrologic facts and terminology. Assuming knowledge of only introductory college-level courses in physics, chemistry, and calculus, it lucidly outlines mathematical derivations fully and at an elementary level, and is ideal for intermediate and advanced courses in igneous and metamorphic petrology. The end-of-chapter quantitative problem sets facilitate student learning by working through simple applications. They also introduce several widely-used thermodynamic software programs for calculating igneous and metamorphic phase equilibria and image analysis software. With over 350 illustrations, this revised edition contains valuable new material on the structure of the Earth's mantle and core, the properties and behaviour of magmas, recent results from satellite imaging, and more.

Introduction to Supercritical Fluids

Phase Equilibria in Ionic Liquid Facilitated Liquid-Liquid Extractions

Fundamentals of Materials Science

Classical and Geometrical Theory of Chemical and Phase Thermodynamics

Understanding the Basics

The only textbook that applies thermodynamics to real-world process engineering problems This must-read for advanced students and professionals alike is the first book to demonstrate how chemical thermodynamics work in the real world by applying them to actual engineering examples. It also discusses the advantages and disadvantages of the particular models and procedures, and explains the most important models that are applied in process industry. All the topics are illustrated with examples that are closely related to practical process simulation problems. At the end of each chapter, additional calculation examples are given to enable readers to extend their comprehension. Chemical Thermodynamics for Process Simulation instructs on the behavior of fluids for pure fluids, describing the main types of equations of state and their abilities. It discusses the various quantities of interest in process simulation, their correlation, and prediction in detail. Chapters look at the important terms for the description of the thermodynamics of mixtures; the most important models and routes for phase equilibrium calculation; models which are applicable to a wide variety of non-electrolyte systems; membrane processes; polymer thermodynamics; enthalpy of reaction; chemical equilibria, and more. -Explains thermodynamic fundamentals used in process simulation with solved examples -Includes new chapters about modern measurement techniques, retrograde condensation, and simultaneous description of chemical equilibrium -Comprises numerous solved examples, which simplify the understanding of the often complex calculation procedures, and discusses advantages and disadvantages of models and procedures -Includes estimation methods for thermophysical properties and phase equilibria thermodynamics of alternative separation processes -Supplemented with MathCAD-sheets and DDBST programs for readers to reproduce the examples Chemical Thermodynamics for Process Simulation is an ideal resource for those working in the fields of process development, process synthesis, or process optimization, and an excellent book for students in the engineering sciences.

Maintaining the substance that made Introduction to the Thermodynamic of Materials a perennial best seller for decades, this Sixth Edition is updated to reflect the broadening field of materials science and engineering. The new edition is reorganized into three major sections to align the book for practical coursework, with the first (Thermodynamic Principles) and second (Phase Equilibria) sections aimed at use in a one semester course, and the third section (Reactions and Transformations) can be used in other courses of the curriculum that deal with oxidation, energy, and phase transformations. The book is updated to include the role of work terms other than PV work (e.g., magnetic work) along with their attendant aspects of entropy, Maxwell equations, and the role of such applied fields on phase diagrams. There is also an increased emphasis on the thermodynamics of phase transformations and the Sixth Edition features an entirely new chapter 15 that links specific thermodynamic applications to the study of phase transformations. The book also features more than 50 new end of chapter problems and more than 50 new figures.

This textbook offers a strong introduction to the fundamental concepts of materials science. It conveys the quintessence of this interdisciplinary field, distinguishing it from merely solid-state physics and solid-state chemistry, using metals as model systems to elucidate the relation between microstructure and materials properties. Mittenmeier's Fundamentals of Materials Science provides a consistent treatment of the subject matter with a special focus on the microstructure-property relationship. Richly illustrated and thoroughly referenced, it is the ideal adoption for an entire undergraduate, and even graduate, course of study in materials science and engineering. It delivers a solid background against which more specialized texts can be studied, covering the necessary breadth of key topics such as crystallography, structure defects, phase equilibria and transformations, diffusion and kinetics, and mechanical properties. The success of the first edition has led to this updated and extended second edition, featuring detailed discussion of electron microscopy, supermicroscopy and diffraction methods, an extended treatment of diffusion in solids, and a separate chapter on phase transformation kinetics. "In a lucid and masterly manner, the ways in which the microstructure can affect a host of basic phenomena in metals are described.... By consistently staying with the postulated topic of the microstructure - property relationship, this book occupies a singular position within the broad spectrum of comparable materials science literature it will also be of permanent value as a reference book for background refreshing, not least because of its unique annotated intermezzo; an ambitious, remarkable work." G. Petzow in International Journal of Materials Research. "The biggest strength of the book is the discussion of the structure-property relationships, which the author has accomplished admirably.... In a nutshell, the book should not be looked at as a quick 'cook book' type text, but as a serious, critical treatise for some significant time to come." G.S. Upadhyaya in Science of Sintering. "The role of lattice defects in deformation processes is clearly illustrated using excellent diagrams. I included are many footnotes, 'Intermezzos', 'Epilogues' and asides within the text from the author's experience. This soon becomes valued for the interesting insights into the subject and shows the human side of its history. Overall this book provides a refreshing treatment of this important subject and should prove a useful addition to the existing text books available to undergraduate and graduate students and researchers in the field of materials science." M. Davies in Materials World.

This textbook provides an intuitive yet mathematically rigorous introduction to the thermodynamics and thermal physics of planetary processes. It demonstrates how the workings of planetary bodies can be understood in depth by reducing them to fundamental physics and chemistry. The book is based on two courses taught by the author for many years at the University of Georgia. It includes 'Guided Exercise' boxes, end-of-chapter problems, and end-of-chapter projects. The third section (Reactions and Transformations) can be used in other courses of the curriculum that deal with oxidation, energy, and phase transformations. The book is updated to include the role of work terms other than PV work (e.g., magnetic work) along with their attendant aspects of entropy, Maxwell equations, and the role of such applied fields on phase diagrams. There is also an increased emphasis on the thermodynamics of phase transformations and the Sixth Edition features an entirely new chapter 15 that links specific thermodynamic applications to the study of phase transformations. The book also features more than 50 new end of chapter problems and more than 50 new figures.

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Thermodynamics of Phase Equilibria in Food Engineering

Stainless Steels for Design Engineers

The Calphad Method

Molecular Thermodynamics of Fluid-phase Equilibria

Basic Principles, Applications, Experimental Techniques

This advanced comprehensive textbook introduces the practical application of phase diagrams to the thermodynamics of materials consisting of several phases. It describes the fundamental physics and thermodynamics as well as experimental methods, treating all material classes: metals, glasses, ceramics, polymers, organic materials, aqueous solutions. With many application examples and realistic cases from chemistry and materials science, it is intended for students and researchers in chemistry, metallurgy, mineralogy, and materials science as well as in engineering and physics. The authors treat the nucleation of phase transitions, the production and stability of technologically important metastable phases, and metallic glasses. Also concisely presented are the thermodynamics and composition of polymer systems. This innovative text puts this powerful analytical approach into a readily understandable and practical context, perhaps for the first time.

Traditionally, the teaching of phase equilibria emphasizes the relationships between the thermodynamic variables of each phase in equilibrium rather than its engineering applications. This book changes the focus from the use of thermodynamics relationships to compute phase equilibria to the design and control of the phase conditions that a process needs. Phase Equilibrium Engineering presents a systematic study and application of phase equilibrium tools to the development of chemical processes. The thermodynamic modeling of mixtures for process development, synthesis, simulation, design and optimization is analyzed. The relation between the mixture molecular properties, the selection of the thermodynamic model and the process technology that could be applied are discussed. A classification of mixtures, separation process, thermodynamic models and technologies is presented to guide the engineer in the world of separation processes. The phase condition required for a given reacting system is studied at subcritical and supercritical conditions. The four cardinal points of phase equilibrium engineering are: the chemical plant or process, the laboratory, the modeling of phase equilibria and the simulator. The harmonization of all these components to obtain a better design or operation is the ultimate goal of phase equilibrium engineering. Methodologies are discussed using relevant industrial examples The molecular nature and composition of the process mixture is given a key role in process decisions Phase equilibrium diagrams are used as a drawing board for process implementation

Phase Equilibrium in Mixtures deals with phase equilibrium and the methods of correlating, checking, and predicting phase data. Topics covered range from latent heat and vapor pressure to dilute solutions, ideal and near-ideal solutions, and consistency tests. Molecular considerations and their use for the prediction and correlation of data are also discussed. Comprised of nine chapters, this volume begins with an introduction to the role of thermodynamics and the criteria for equilibrium between phases, along with fugacity and the thermodynamic functions of mixing. The discussion then turns to some of the phase phenomena which may be encountered in chemical engineering practice; methods of correlating and extending vapor pressure data and practical techniques for calculating latent heats from these data; the behavior of dilute solutions both at low and high pressures for reacting and non-reacting systems; and the behavior of ideal and near-ideal solutions. The remaining chapters explore non-ideal solutions at normal pressures; practical methods for testing the thermodynamic consistency of phase data; and the extent to which the broad aspects of phase behavior may be interpreted in the light of simple molecular considerations. This book is intended primarily for graduate chemical engineers but should also be of interest to those graduates in physics or chemistry who need to use phase equilibrium data.

Written by a leading practitioner and teacher in the field of ceramic science and engineering, this outstanding text provides advanced undergraduate- and graduate-level students with a comprehensive, up-to-date Introduction to Phase Equilibria in Ceramic Systems Building upon a concise definition of the phase rule, the book logically proceeds from one- and two-component systems through increasingly complex systems, enabling students to utilize the phase rule in real applications. Unique because of its emphasis on phase diagrams, timely because of the rising importance of ceramic applications, practical because of its pedagogical approach, Introduction to Phase Equilibria in Ceramic Systems offers end-of-chapter review problems, ... extending lists ... a solid thermodynamic foundation, ... and clear perspectives on the special properties of ceramics as compared to metals. This authoritative volume fills a broad gap in the literature, helping undergraduate- and graduate-level students of ceramic engineering and materials science to approach this demanding subject in a rational, confident fashion. In addition, Introduction to Phase Equilibria in Ceramic Systems serves as a valuable supplement to undergraduate-level metallurgy programs.

Teaching Mineralogy

Introduction to Phase Equilibria in Ceramic Systems

Alloy Phase Equilibria

Introduction to the Thermodynamics of Materials

CRC Handbook of Phase Equilibria and Thermodynamic Data of Copolymer Solutions

This monograph acts as a benchmark to current achievements in the field of Computer Coupling of Phase Diagrams and Thermochemistry, often called CALPHAD which is an acronym for Computer Calculation of PHase Diagrams. It also acts as a guide to both the basic background of the subject area and the cutting edge of the topic, combining comprehensive discussions of the underlying physical principles of the CALPHAD method with detailed descriptions of their application to real complex multi-component materials. Approaches which combine both thermodynamic and kinetic models to interpret non-equilibrium phase transformations are also reviewed.

Phase Equilibria: Basic Principles, Applications, Experimental Techniques presents an analytical treatment in the study of the theories and principles of phase equilibria. The book is organized to afford a deep and thorough understanding of such subjects as the method of species model systems; condensed phase-vapor phase equilibria and vapor transport reactions; zone refining techniques; and nonstoichiometry. Physicists, physical chemists, engineers, and materials scientists will find the book a good reference material.

This volume presents a sound foundation for understanding abstract concepts (physical properties such as fugacity, or chemical processes, such as distillation) of phase and reaction equilibria, and shows you how to apply these concepts to solve practical problems using numerous, clear examples. The book encourages the use of MATHCAD to write programs specific to each problem, enabling you to easily track mistakes and understand the order of magnitude of the various quantities involved. Provides guidelines in order to choose the 'best' equation of state suitable for the particular situation Includes up-to-date information, comprehensive in-depth content and current examples in each chapter Provides the right tools in order to and encourages you to use MATHCAD to write your own specific programs Includes many well organized problems (real solutions), which are extensions of the examples enabling conceptual understanding to quantitative/real problem solving Includes all mathematical background required for solving problems encountered in phase and reaction equilibria Provides a Solutions Manual (for instructors in pdf form) allowing the use of the book in advanced thermodynamic courses

This new book provides, for the first time, a thorough survey of the techniques and equipment for both high- and low-pressure phase equilibrium measurement and addresses the equally challenging task of accurately modeling or predicting the equilibria. The book is unique because it combines in depth and authoritative coverage of both experimental and theoretical procedures in a single volume. Written as a reference for practicing engineers and scientists in the chemical engineering field, this book will also be useful as an advanced graduate-level text.

Phenomenology and Computation

Phase Diagrams and Thermodynamic Modeling of Solutions

Encyclopedia of Geochemistry

Phase Equilibrium Engineering

Phase Equilibrium in Mixtures

Thermodynamics of Phase Equilibria in Food Engineering is the definitive book on thermodynamics of equilibrium applied to food engineering. Food is a complex matrix consisting of different groups of compounds divided into macronutrients (lipids, carbohydrates, and proteins), and micronutrients (vitamins, minerals, and phytochemicals). The quality characteristics of the sensorial, physical and microbiological attributes are directly related to the thermodynamic properties of specific compounds and complexes that are formed during processing or by the action of diverse interventions, such as the environment, biochemical reactions, and others. In addition, in obtaining bioactive substances using separation processes, the knowledge of phase equilibria is essential to provide an efficient separation, with a low cost in the process and high selectivity in the recovery of the desired component. This book combines theory and application of phase equilibria data of systems containing food compounds to help food engineers and researchers to solve complex problems found in food processing. It provides support and industry to better understand the behavior of food materials in the face of processing effects, and to develop ways to improve the quality of the food products. Presents the fundamentals of phase equilibria in the food industry Describes both classic and advanced models, including cubic equations of state and activity coefficient Encompasses distillation, solid extraction, adsorption, crystallization and supercritical fluid extraction Explores equilibrium in advanced systems, including colloidal, electrolyte and protein systems

Phase Diagrams and Thermodynamic Modeling of Solutions provides readers with an understanding of thermodynamics and phase equilibria that is required to make full and efficient use of these tools. The book systematically discusses phase diagrams of all types, the thermodynamics behind them, their calculations from thermodynamic databases, and the structure of the development of these databases. Featuring examples from a wide range of systems including metals, salts, ceramics, refractories, and concentrated aqueous solutions, Phase Diagrams and Thermodynamic Modeling of Solutions is a vital resource for researchers and developers in materials science, metallurgy, combustion and energy, corrosion engineering, environmental engineering, glass technology, nuclear engineering, and other fields of inorganic chemical and materials science and engineering. Additionally, experts involved in developing thermodynamic databases will find a comprehensive reference text of current solution models. Presents a rigorous and complete development of thermodynamics for readers who already have a basic understanding of thermodynamics Provides an in-depth understanding of phase equilibria Includes information that can be used as a text for graduate courses on thermodynamics and phase diagrams, or on solution modeling Covers several types of phase diagrams (paraequilibrium, solidus projections, first-melting projections, Schell diagrams, enthalpy diagrams), and more Phase diagrams are used in materials research and engineering to understand the interrelationship between composition, microstructure and process conditions. In complex systems, computational methods such as CALPHAD are employed to model thermodynamic properties for each phase and simulate multicomponent phase behavior. Written by recognized experts in the field, this book provides a comprehensive, up-to-date Introduction to Phase Equilibria in Ceramic Systems Building upon a concise definition of the phase rule, the book logically proceeds from one- and two-component systems through increasingly complex systems, enabling students to utilize the phase rule in real applications. Unique because of its emphasis on phase diagrams, timely because of the rising importance of ceramic applications, practical because of its pedagogical approach, Introduction to Phase Equilibria in Ceramic Systems offers end-of-chapter review problems, extensive reading lists, a solid thermodynamic foundation and clear perspectives on the special properties of ceramics as compared to metals. This authoritative volume fills a broad gap in the literature, helping undergraduate- and graduate-level students of ceramic engineering and materials science to approach this demanding subject in a rational, confident fashion. In addition, Introduction to Phase Equilibria in Ceramic Systems serves as a valuable supplement to undergraduate-level metallurgy programs.

Because it is grounded in math, chemical thermodynamics is often perceived as a difficult subject and many students are never fully comfortable with it. The first authoritative textbook presentation of equilibrium chemical and phase thermodynamics in a reformulated geometrical framework, Chemical and Phase Thermodynamics shows how this famously difficult subject can be accurately expressed with only elementary high-school geometry concepts. Featuring numerous suggestions for research-level extensions, this simplified alternative to standard calculus-based thermodynamics expositions is perfect for undergraduate and beginning graduate students as well as researchers.

Introduction to Phase Equilibria in Ceramics

International Series of Monographs in Chemical Engineering

Chemical Thermodynamics for Process Simulation

Computational Thermodynamics

Phenomenology and Thermodynamics